Impact of the support on the activity and stability of Ir catalysts under the Dry Reforming of Methane conditions



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1.INTRODUCTION

Simultaneous use of the two most important greenhouse gases, CO₂ and CH_4 , providing an effective method for sustainable carbon recycling through a more environmentally friendly approach to natural gas exploitation, as well as direct utilization of biogas [1].

DRM: $CH_4 + CO_2 \rightarrow 2H_2 + 2CO$, $\Delta H^{\circ}_{298} = 247$ kJ/mol

Within the temperature range of 650-800°C, intermediate reactions occur which affect both the H_2/CO ratio of the produced synthesis gas and the cumulative deposition of carbon, with the latter being one of the major problems in many catalytic systems. The two main reactions that lead to carbon deposition are :

2.EXPERIMENTAL

2.1 Synthesis and characterization

✓ The supports of mixed oxides , ACZ (80%wt γ -Al₂O₃ -20%wt Ce_{0.5}Zr_{0.5}O_{2-δ}) and CZ (Ce_{0.5}Zr_{0.5}O_{2-δ}) were prepared by the co-precipitation method, while the alumina oxide γ -Al₂O₃ was sourced from commerce.

 \checkmark The supported Ir catalysts were prepared by the wet impregnation method, with a targeted metal loading of ~1wt%. After wet impregnation, the catalysts were reduced under continuous flow of 50% H_2/He (50ml/min) at 400°C for 2h, followed by a flow of 1% H_2/He (50ml/min) at 850°C for 1h (referred to as "fresh"). The exact determination of Ir loading on the catalysts was performed by ICP-OES (1wt%) Ir/Al_2O_3 , 0.4wt% Ir/ACZ, and 0.6wt% Ir/CZ).

 \checkmark The structural and morphological characteristics of the catalysts, as well as their other physicochemical properties, were determined using BET, XRD, HRTEM, H₂-TPR, H₂-chemisorption,

Boudouard reaction	Methane cracking	
$2CO \rightleftharpoons CO_2 + C$ $\Delta H^\circ = -172 \text{ kJ/mol}$	$CH_4 \rightleftharpoons 2H_2 + C$ $\Delta H^\circ = +75 \text{ kJ/ mol}$	

In this study, the effect of support on the DRM performance (activity at 500-750°C) of Ir nanoparticles, carbon deposition phenomena, and their stability under reaction (at 750°C) and/or after oxidative thermal aging conditions is investigated. Supports studied include γ-Al₂O₃, Alumina-Ceria-Zirconia (ACZ) and Ceria-Zirconia (CZ).

3.RESULTS & DISCUSSION

3.1 Activity and stability experiments

Table 1: Textural, morphological and reducibility characteristics of the -Al₂O₃, ACZ and CZ supports and the corresponding as-prepared (fresh) Ir/-Al₂O₃, Ir/ACZ and Ir/CZ catalysts.(BET, H₂TPR, H₂ chemisorption results)

Supports/	S _{BET}	Total Pore	OSC	Dispersion	Mean Ir particle
Catalysts	(cm²/g)	Volume (cm ³ /g)	(µmol O ₂ /g)	D _{lr} (H/Ir)	size (nm)
γ-Al ₂ O ₃	178	0.6	0	-	-
Ir/γ-Al ₂ O ₃	167	0.57	38	0.7	1.2
ACZ	149	0.29	110	-	-
Ir/ACZ	73	0.22	176	0.41	1.8
CZ	22	0.05	557	-	_

ICP-OES and TPO-carbon deposition studies

2.2 Catalytic experiments (activity and stability experiments/ oxidative thermal aging)

 \checkmark Catalytic experiments were carried out in an isothermal fixed-bed reactor (quartz i.d.=3mm) with a catalyst loading of 50 mg. The reactor feed composition consisted of 50% CH₄ and 50% CO₂ simulating biomass gasification. The total feed flow rate (F_t) was 100 mL/min (WGHSV = 120,000 mL/gcat·h). The performance of the catalysts was studied in the temperature range of 500-750°C and their stability at 750°C for 12h.

 \checkmark After evaluating the fresh catalysts in the DRM reaction, we proceeded with their oxidative aging, which was performed in two successive steps: (i) 2h oxidation under 20% O₂ flow in He (50ml/min) at 650° C (aged@650), and (ii) an additional 2h oxidation under 20% O₂ flow in He (50ml/min) at 750°C (aged@750). After each step, the catalysts were re-evaluated in the DRM under constant temperature conditions (750°C) and a feed of CH_4 :CO₂=50%:50% with a flow of 100 mL/min (GHSV = 120,000) $mL/g \cdot h$) and pressure of 1 bar.



Scheme 1. Graphical representation of the DRM reaction pathways over Ir nanoparticles dispersed on Al_2O_3 and CZ-containing supports. The promotional effects induced by CZ-containing supports with high OSC values originate from the presence of the surface oxygen vacancies, labile O²⁻ species and back-spillover phenomena.



Figure 4. Temperature programmed oxidation





Figure 1. H₂-TPR profiles of the Al₂O₃, ACZ and CZ supports (a), and the Ir/-Al₂O₃, Ir/ACZ and Ir/CZ catalysts (b)

Figure 3. Arrhenius plots showing the temperature dependence of (a) the CH_4 (closed symbols) and

3.2 Oxidative thermal aging



Table 3. Textural and morphological characteristics of the "aged@750" catalysts.

Catalysts	S _{BET}	OSC	Mean Ir particle	Mean Ir particle
	cm²/g	support	size (nm) of	size (nm) of
		µmolO ₂ /g	fresh catalysts	aged catalysts
Ir/γ-Al ₂ O ₃	140	0	1.2	13.6
Ir/γ-Al ₂ O ₃ Ir/ACZ	140 64	0 110	1.2 1.8	13.6 2.1



profiles of fresh catalysts Ir/γ -Al₂O₃, Ir/ACZ, and Ir/CZ after 3 hours of operation under DRM reaction conditions (CH₄:CO₂=50%, F=100 mL/min, T=750°C)

4.CONCLUSIONS

✓The use of ACZ and CZ supports with sufficient OSC and surface oxygen vacancies significantly enhances the CO₂ consumption rate while not affecting the CH₄ consumption rate. This implies the occurrence of a bifunctional reaction mechanism.

✓The catalysts exhibited excellent timeon-stream stability due to the low propensity of Ir to form and accumulate carbon, as well as the resistance of metallic Ir phase to agglomeration.

✓The Ir/ACZ and mainly Ir/CZ catalysts show a lower tendency for carbon accumulation, verifying the important role of labile O₂ species of the support on the gasification rate of surface carbon species. ✓ Ir nanoparticles deposited on supports with a high OSC value display remarkable stability (resistance to agglomeration) under conditions of intense oxidative thermal aging.



Time-on-stream (h)

Figure2. CH₄ and CO₂ consumption rates and H_2/CO molar ratios (inset) obtained with the fresh Ir/Al₂O₃, Ir/ACZ and Ir/CZ catalysts as a function of time-on-stream . Experimental conditions :T = 750°C, $[CH_4]in = [CO_2]in = 50\%$,1 bar, WGHSV = 120,000 mL/gcat·h , w_{cat} = 50 CO₂ (open symbols) consumption rates, and (b) the corresponding H_2/CO molar ratios, obtained for fresh Ir/Al₂O₃, Ir/ACZ and Ir/CZ catalysts. Experimental conditions: $([CH_4] = [CO_2] = 50 \%, 1$ bar, $w_{cat} = 50$ mg, WGHSV = 120,000–200,000 mL/gcat·h.
 Table 2.
 Support-induced
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 apparent
activation energies during DRM over Ir/Al_2O_3 , Ir/ACZ and Ir/CZ catalysts.

Catalysts	Support OSC (µmol O ₂ /g)	Activation Energy (Ea) (KJ/mol)
Ir/γ-Al ₂ O ₃	0	91.3
Ir/ACZ	110	88.5
Ir/CZ	557	73.5

(b) and H_2/CO molar ratio (c), of Ir/γ -Al₂O₃, Ir/ACZ and Ir/CZcatalysts. Aging protocols: (i) Aged@650 (2h in situ oxidation with 50 mL/min flow of 20% O₂/He at 650°C) (ii) Aged@750 (consecutive 2h additional in situ oxidation with 50 mL/min flow of 20% O_2 /He at 750°C). Experimental conditions:750°C , $[CH_4] = [CO_2] = 50\%$,1 bar, mcat =50mg, WGHSV = 120,000 mL/gcat·h. DRM performance values given are the mean values of 5-6 measurements taken in a period of \sim 6 h of stable time - on - stream operation.

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5.REFERENCES

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Acknowledgments

The research project is implemented in the framework of H.F.R.I call "Basic research Financing (Horizontal support of all Sciences)" under the National Recovery and Resilience Plan "Greece 2.0" funded by the European Union – NextGenerationEU (H.F.R.I. Project Number:16916).

